error of the means are reported where appropriate.
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# Structure-Activity Studies of Trichothecenes: Cytotoxicity of Analogues and Reaction Products Derived from T-2 Toxin and Neosolaniol 

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#### Abstract

Forty-two analogues and reaction products derived from T-2 toxin or neosolaniol were assayed for their cytotoxicity to cultured mouse lymphoma cells. Structure-activity relationships confirmed the stereospecific nature of the cytotoxic action of T-2. Cytotoxicity was particularly susceptible to changes at $\mathrm{C} 3, \mathrm{C} 4, \mathrm{C} 9$, and C 10 but was relatively unaffected by changes at C 8 , which appears to represent a region of steric tolerance in the interaction of T-2 with a cellular constituent. The most potent compounds were T-2, diacetoxyscirpenol, and a series of C8 ester analogues 11 and 31-35.


Trichothecene mycotoxins are toxic secondary metabolites produced by various species of fungi such as Fusarium, Trichoderma, Trichothecium, Myrothecium, Stachybotrys, Cephalosporium, and Verticimonosporium. ${ }^{1}$ Chemically they are sesquiterpenoids, usually containing a relatively unreactive epoxide moiety. Trichothecenes, particularly those produced by species of Fusarium such as deoxynivalenol and T-2 toxin (1), have been implicated as the causative agents of various mycotoxicoses occurring in farm animals and man. ${ }^{2}$ At the cellular level trichothecenes are potent inhibitors of protein synthesis in eukaryotic cells, ${ }^{3}$ and one of the more potent naturally occurring metabolites, diacetoxyscirpenol (2) (DAS, anguidine), has undergone clinical trials as an anticancer agent. ${ }^{4}$


$$
\begin{aligned}
& R=O=O C \mathrm{OH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \\
& R=H
\end{aligned}
$$

Structure-activity relationships of trichothecenes are therefore of interest with regard to the possible development of novel anticancer agents and to the etiology of mycotoxicoses occurring in animals and man. The first major structure-activity study was reported by Grove and Mortimer ${ }^{5}$ following the isolation and characterization of diacetoxyscirpenol. Eighteen analogues and rearrangement products derived from diacetoxyscirpenol and nivalenol were assayed for their cytotoxicity toward HeLa cells. The main conclusion was that destruction of the epoxide moiety
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and rearrangement of the ring system virtually eliminated cytotoxicity. Further structure-activity studies, predominantly using trichothecenes isolated from laboratory cultures, have been reported from several laboratories. ${ }^{6-11}$ More recently approximately 60 synthetic analogues derived from diacetoxyscirpenol were evaluated for their antitumor properties in mice. ${ }^{12}$ Jarvis and co-workers have reported extensive investigations of the more complex macrocyclic trichothecenes. ${ }^{13}$ All of these studies indicate a stereospecific interaction of trichothecenes with a receptor, presumed to be located on the ribosome. In this present paper we report additional structure-activity studies of synthetic trichothecenes, derived by modification of one of the most potent of the naturally occurring trichothecenes, T-2 toxin, or from its partial hydrolysis product neosolaniol (3). Some of the compounds reported herein were synthesized systematically as part of the structure-activity study; others were products of chemical investigations of T-2 toxin.

## Chemistry

The natural metabolites T-2 toxin and neosolaniol were used as starting materials for this work. Analogues modified at C3, C4, or C15 are shown in Table I, those modified at C8 are shown in Table II, and those modified at C9-C10 are shown in Table III. New derivatives were characterized principally by NMR and mass spectroscopy. NMR as-
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Table I. Structures of T-2 and Analogues Modified at C3, C4, or C15 and Their Cytotoxicity to Mouse Lymphoma Cells

${ }^{a}$ Mouse lymphoma cells L5178Y. ${ }^{b}$ i-Val $=\mathrm{COCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} .{ }^{c} \mathrm{Ms}=\mathrm{SO}_{2} \mathrm{Me}$. ${ }^{d} \mathrm{TBDMS}=\mathrm{Si}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}$.

Table II. Structures of T-2 Analogues Modified at C8 and Their Cytotoxicity to Mouse Lymphoma Cells


| compd | $\mathrm{R}_{\mathbf{6}}$ | $\mathrm{R}_{7}$ | cytotoxicity: <br> $\mathrm{LC}_{50}, \mathrm{ng} / \mathrm{mL}$ |
| :---: | :--- | :--- | :---: |
| $\mathbf{2}$ | H | H | 2 |
| $\mathbf{3}$ | OH | H | 40 |
| 11 | OAc | H | 1.4 |
| $\mathbf{3 1}$ | OCOEt | H | 1.6 |
| $\mathbf{3 2}$ | OCOPr | H | 1.2 |
| $\mathbf{3 3}$ | OCOBu | H | 1.0 |
| $\mathbf{3 4}$ | OCOPe | H | 1.3 |
| $\mathbf{3 5}$ | OCOPh | H | 2.1 |
| $\mathbf{3 6}$ |  | $=\mathrm{O}$ |  |
| $\mathbf{3 7}$ | H | OH | 6.9 |

signments were based on correlations that have been well established. ${ }^{12,14,15}$ Selected ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data are shown in Tables IV and V. Elemental analyses were

[^0]Table III. Structures of T-2 Analogues Modified at C9-C10 and Their Cytotoxicity to Mouse Lymphoma Cells

| compd | $\mathrm{R}_{1}$ | $\mathrm{R}_{4}$ | $\mathrm{R}_{6}$ | $\mathrm{R}_{8}$ | $\mathrm{R}_{6}$ | cytotoxicity: <br> $\mathrm{LC}_{60}, \mathrm{ng} / \mathrm{mL}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  |  |  |
| 38 |  |  | $\mathrm{Oi}-\mathrm{Va}$ | $\mathrm{H}^{\text {a }}$ | H | 30 |
| 39 |  |  | H | $\mathrm{H}^{\text {a }}$ | H | 50 |
| 40 |  |  | $\mathrm{Oi}-\mathrm{Va}$ |  |  | 500 |
| 43 |  |  | $\mathrm{Oi}-\mathrm{Val}$ | OH | OH | 85 |
|  |  |  |  |  |  |  |
| 41 | OH | OH | OH |  | OH | 2000 |
| 42 | OAc | OAc | OAc |  | OAc | 600 |
| 45 | OH | OH | OH |  | Br | 50000 |
| 46 | OAc | OAc | OAc |  | Br | 50000 |

[^1]obtained for most new compounds; those isolated as solid foams or oils for which they were not obtained were shown to be $>95 \%$ pure by ${ }^{13} \mathrm{C}$ NMR and TLC.
Isolation of T-2 and Neosolaniol. T-2 toxin and neosolaniol were isolated from cultures of $F$. sporotrichioides strain NRRL 3299 on rice, using culture condi-

Table IV. ${ }^{1} \mathrm{H}$ NMR Spectral Data ( 100 MHz ) for $\mathrm{T} \cdot 2$ Derivatives, Chemical Shifts, and Multiplicity ( $J$ in hertz)

|  | position |  |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| compd | 2 | 3 | 4 | 7 | 8 | 10 | 11 | 13 | 14 | 15 | 16 |
| 8 | $\begin{aligned} & 3.75 \mathrm{~d} \\ & (5.0) \end{aligned}$ | $\begin{aligned} & 5.4 \mathrm{dd} \\ & (6.6) \end{aligned}$ | $\begin{gathered} 6.1 \mathrm{~d} \\ (6.6) \end{gathered}$ | $\begin{aligned} & 1.8,2.4 \mathrm{ABX} \\ & (5.5,14.8) \end{aligned}$ | $\begin{aligned} & 5.25 \mathrm{ABX} \\ & (5.5) \end{aligned}$ | $\begin{aligned} & 5.7 \mathrm{~m} \\ & (5.5) \end{aligned}$ | $\begin{aligned} & 4.05 \mathrm{~d} \\ & (5.5) \end{aligned}$ | $\begin{aligned} & 2.8,3.1 \mathrm{AB} \\ & (3.9) \end{aligned}$ | 0.8 s | $\underset{(14.4)}{4.1,4.3 \mathrm{AB}}$ | $1.75 \mathrm{~s}$ |
| 10 | $\begin{gathered} 3.8 \mathrm{~d} \\ (5.0) \end{gathered}$ | $5.15 \mathrm{dd}$ | $\begin{gathered} 5.85 \mathrm{~d} \\ (3.0) \end{gathered}$ | $\begin{aligned} & 1.8,2.6 \mathrm{ABX} \\ & (5.5,14.0) \end{aligned}$ | 4.1 ABX | $\begin{aligned} & 5.6 \mathrm{~m} \\ & (5.5) \end{aligned}$ | 4.1 d | $\begin{aligned} & 2.8,3.05 \mathrm{AB} \\ & (4.0) \end{aligned}$ | 0.75 s | 4.3 AB | $1.85 \mathrm{~s}$ |
| 11 | $\begin{gathered} 3.7 \mathrm{~d} \\ (4.8) \end{gathered}$ | $\begin{gathered} 4.2 \mathrm{dd} \\ (2.3,4.8) \end{gathered}$ | $\begin{gathered} 5.2 \mathrm{~d} \\ (2.3) \end{gathered}$ | $\begin{aligned} & 2.0,2.35 \mathrm{ABX} \\ & (5.6,12.9) \end{aligned}$ | $\begin{aligned} & 5.25 \mathrm{ABX} \\ & (5.0) \end{aligned}$ | $\begin{aligned} & 5.8 \mathrm{~m} \\ & (5.8) \end{aligned}$ | $\begin{gathered} 4.2 \mathrm{~d} \\ (5.8) \end{gathered}$ | $\begin{aligned} & 2.8,3.05 \mathrm{AB} \\ & (4.0) \end{aligned}$ | 0.8 s | $\begin{aligned} & \text { 4.1, 4.3 AB } \\ & (12.2) \end{aligned}$ | $\begin{aligned} & 1.75 \mathrm{~s} \\ & - \end{aligned}$ |
| 12 | $\begin{gathered} 3.85 \mathrm{~d} \\ (5.4) \end{gathered}$ | $\begin{gathered} 5.15 \mathrm{dd} \\ (3.2,4.6) \end{gathered}$ | $\begin{aligned} & 5.85 \mathrm{~d} \\ & (3.2) \end{aligned}$ | $\begin{aligned} & \text { 2.0, 2.3 ABX } \\ & (5.7,15.0) \end{aligned}$ | $\begin{aligned} & 5.25 \mathrm{ABX} \\ & (5.7) \end{aligned}$ | $\begin{aligned} & 5.75 \mathrm{~m} \\ & (5.7) \end{aligned}$ | $\begin{aligned} & 4.15 \mathrm{~d} \\ & (5.7) \end{aligned}$ | $\begin{aligned} & 2.8,3.05 \mathrm{AB} \\ & (3.8) \end{aligned}$ | 0.75 s | $4.1,4.35 \mathrm{AB}$ | $1.75 \mathrm{~s}$ |
| 13 | $\begin{gathered} 3.8 \mathrm{~d} \\ (5.0) \end{gathered}$ | $\begin{aligned} & 2.55 \mathrm{~m} \\ & (7.5,12.9) \end{aligned}$ | $\begin{aligned} & 5.95 \mathrm{~d} \\ & (7.6) \end{aligned}$ | $\begin{aligned} & 2.35 \mathrm{ABX} \\ & (5.5,15.0) \end{aligned}$ | $\begin{aligned} & 5.25 \mathrm{ABX} \\ & (5.5) \end{aligned}$ | $\begin{aligned} & 5.7 \mathrm{~m} \\ & (5.6) \end{aligned}$ | $\begin{aligned} & 3.96 \mathrm{~d} \\ & (5.6) \end{aligned}$ | $\begin{aligned} & 2.85,3.1 \mathrm{AB} \\ & (3.6) \end{aligned}$ | $0.8 \mathrm{~s}$ | $\begin{aligned} & \text { 4.3, 4.6 AB } \\ & (12.0) \end{aligned}$ | $1.7 \mathrm{~s}$ |
| 14 | $\begin{gathered} 3.8 \mathrm{~d} \\ (5.0) \end{gathered}$ | $\begin{aligned} & 2.6 \mathrm{~m} \\ & (7.2,15.4) \end{aligned}$ | 4.6 d | $\begin{gathered} 2.3 \mathrm{ABX} \\ (5.2,14.8) \end{gathered}$ | $\begin{aligned} & 5.25 \mathrm{ABX} \\ & (5.2) \end{aligned}$ | $\begin{aligned} & 5.65 \mathrm{~m} \\ & (5.6) \end{aligned}$ | $\begin{gathered} 3.8 \mathrm{~d} \\ (5.6) \end{gathered}$ | $\begin{aligned} & 2.8,3.1 \mathrm{AB} \\ & (3.6) \end{aligned}$ | $0.85 \mathrm{~s}$ | $\begin{aligned} & 4.0,4.25 \mathrm{AB} \\ & (12.4) \end{aligned}$ | $1.7 \mathrm{~s}$ |
| 15 | $\begin{gathered} 3.75 \mathrm{~d} \\ (5.0) \end{gathered}$ | $\begin{aligned} & 2.5 \mathrm{~m} \\ & (7.2,15.2) \end{aligned}$ | $4.35 \mathrm{~d}$ | 2.1 ABX | - - 5 | $\begin{aligned} & 5.55 \mathrm{~m} \\ & (5.5) \end{aligned}$ | , | $\begin{aligned} & 2.8,3.1 \mathrm{AB} \\ & (3.9) \end{aligned}$ | $0.95 \mathrm{~s}$ | , | $1.85 \mathrm{~s}$ |
| 16 | $\begin{gathered} 5.0 \mathrm{~d} \\ (5.0) \end{gathered}$ |  | 6.5 s | $\begin{aligned} & 1.75,2.35 \mathrm{ABX} \\ & (6.4,15.0) \end{aligned}$ | $\begin{aligned} & \text { 5.3 ABX } \\ & - \end{aligned}$ | $\begin{aligned} & 5.7 \mathrm{~m} \\ & (6.0) \end{aligned}$ | $\begin{aligned} & 4.45 \mathrm{~d} \\ & (6.0) \end{aligned}$ | $\begin{aligned} & 2.9,2.95 \mathrm{AB} \\ & (3.7) \end{aligned}$ | $0.8 \mathrm{~s}$ | $\begin{aligned} & 4.1,4.4 \mathrm{AB} \\ & (14.5) \end{aligned}$ | $1.75 \mathrm{~s}$ |
| 20 | $\begin{aligned} & 3.75 \mathrm{~d} \\ & (4.5) \end{aligned}$ | $\begin{gathered} 4.8 \mathrm{dd} \\ (3.0,4.5) \end{gathered}$ | $\begin{gathered} 4.4 \mathrm{~d} \\ (3.0) \end{gathered}$ | $\begin{aligned} & \text { 2.0, 2.3 ABX } \\ & (5.0,15.0) \end{aligned}$ | $\begin{aligned} & 5.25 \mathrm{ABX} \\ & (5.0) \end{aligned}$ | $\begin{aligned} & 5.7 \mathrm{~m} \\ & (5.5) \end{aligned}$ | $\begin{aligned} & 3.85 \mathrm{~d} \\ & (5.5) \end{aligned}$ | $\begin{aligned} & 2.75,3.0 \mathrm{AB} \\ & (4.0) \end{aligned}$ | 0.8 s - | $\begin{aligned} & 3.95,4.2 \mathrm{AB} \\ & (12.0) \end{aligned}$ | $1.7 \mathrm{~s}$ |
| 21 | $\begin{aligned} & 4.15 \mathrm{~d} \\ & (4.8) \end{aligned}$ | $\begin{aligned} & 5.3 \mathrm{~d} \\ & (4.8) \end{aligned}$ | - | $\begin{aligned} & 2.0,2.4 \mathrm{ABX} \\ & (4.4,14.6) \end{aligned}$ | $\begin{aligned} & \text { 5.3 ABX } \\ & (4.8) \end{aligned}$ | $\begin{aligned} & 5.7 \mathrm{~m} \\ & (5.0) \end{aligned}$ | $\begin{gathered} 4.0 \mathrm{~d} \\ (5.0) \end{gathered}$ | $\begin{aligned} & 3.0,3.25 \mathrm{AB} \\ & (3.8) \end{aligned}$ | $0.9 \mathrm{~s}$ | $\begin{aligned} & 3.95,4.05 \mathrm{AB} \\ & (12.5) \end{aligned}$ | $1.75 \mathrm{~s}$ |
| 24 | $\begin{gathered} 3.75 \mathrm{~d} \\ (4.5) \end{gathered}$ | $\begin{gathered} 5.15 \mathrm{dd} \\ (5.2,9.5) \end{gathered}$ | $\begin{aligned} & 2.0 \mathrm{~m} \\ & (4.5,9.5) \end{aligned}$ | - | $\begin{aligned} & \text { 5.3 ABX } \\ & (5.0) \end{aligned}$ | $\begin{aligned} & 5.75 \mathrm{~m} \\ & (5.5) \end{aligned}$ | $\begin{gathered} 4.2 \mathrm{~d} \\ (5.5) \end{gathered}$ | $\begin{aligned} & 2.85,3.1 \mathrm{AB} \\ & (3.8) \end{aligned}$ | $0.8 \mathrm{~s}$ | $\begin{aligned} & 3.95,4.2 \mathrm{AB} \\ & (12.0) \end{aligned}$ | $1.75 \mathrm{~s}$ |
| 25 | $\begin{aligned} & 3.7 \mathrm{~d} \\ & (3.6) \end{aligned}$ |  | - | - | $\begin{aligned} & \text { 5.3 ABX } \\ & (5.4) \end{aligned}$ | $\begin{aligned} & 5.7 \mathrm{~m} \\ & (5.6) \end{aligned}$ | $\begin{aligned} & 3.9 \mathrm{~d} \\ & (5.6) \end{aligned}$ | $\begin{aligned} & 2.9,3.15 \mathrm{AB} \\ & (3.6) \end{aligned}$ | $0.8 \mathrm{~s}$ | $\begin{aligned} & 3.95,4.2 \mathrm{AB} \\ & (11.8) \end{aligned}$ | $1.75 \mathrm{~s}$ |
| 26 | $\begin{gathered} 4.01 \mathrm{dd} \\ (0.8,2.7) \end{gathered}$ | $\begin{gathered} 6.1 \mathrm{dd} \\ (2.7,5.9) \end{gathered}$ | $\begin{gathered} 6.4 \mathrm{dd} \\ (0.8,5.9) \end{gathered}$ | $\begin{gathered} \text { 2.3 ABX } \\ (5.1,14.8) \end{gathered}$ | $\begin{aligned} & 5.35 \mathrm{ABX} \\ & (5.1) \end{aligned}$ | $\begin{aligned} & 5.65 \mathrm{~m} \\ & (5.1) \end{aligned}$ | $\begin{aligned} & 4.05 \mathrm{~d} \\ & (5.1) \end{aligned}$ | $\begin{aligned} & 3.0,3.2 \mathrm{AB} \\ & (3.6) \end{aligned}$ | 0.95 s | $\begin{aligned} & 3.85,4.15 \mathrm{AB} \\ & (11.9) \end{aligned}$ | $1.75 \mathrm{~s}$ |
| 28 | 3.35 s |  | $\begin{aligned} & 2.3,2.7 \mathrm{~d} \\ & (19.0) \end{aligned}$ | $\begin{aligned} & \text { 2.2 ABX } \\ & (5.0) \end{aligned}$ | 4.15 ABX | $\begin{gathered} 6.55 \mathrm{~m} \\ (1.2,5.5) \end{gathered}$ | 4.0 d | $\begin{aligned} & 3.2,3.55 \mathrm{AB} \\ & (4.0) \end{aligned}$ | 1.05 s | 3.6 AB | 1.95 s |
| 29 | 3.4 s | - | $\begin{aligned} & 3.0,6.3 \mathrm{~d} \\ & (18.5) \end{aligned}$ | $\begin{aligned} & 2.0,2.4 \mathrm{ABX} \\ & (5.5,14.7) \end{aligned}$ | $\begin{aligned} & \text { 5.3 ABX } \\ & (4.5) \end{aligned}$ | $\begin{aligned} & 5.7 \mathrm{~m} \\ & (5.5) \end{aligned}$ | $\underset{(5.5)}{4.1 \mathrm{~d}}$ | $\begin{aligned} & 3.1,3.25 \mathrm{AB} \\ & (3.8) \end{aligned}$ | 1.0 s | $\begin{aligned} & 4.0,4.2 \mathrm{AB} \\ & (12.0) \end{aligned}$ | $1.75 \mathrm{~s}$ |
| 30 | $\begin{gathered} 3.5 \mathrm{~d} \\ (4.6) \end{gathered}$ | $4.3 \mathrm{dd}$ | $\begin{aligned} & 5.7 \mathrm{~d} \\ & (2.6) \end{aligned}$ | $\begin{aligned} & 1.8,2.3 \mathrm{ABX} \\ & (5.3,14.2) \end{aligned}$ | $\begin{aligned} & \text { 4.1 ABX } \\ & (5.3) \end{aligned}$ | $\begin{aligned} & 5.6 \mathrm{~m} \\ & (5.5) \end{aligned}$ | $\begin{aligned} & 4.3 \mathrm{~d} \\ & (5.5) \end{aligned}$ | $\begin{aligned} & 2.75,3.0 \mathrm{AB} \\ & (4.0) \end{aligned}$ | 0.7 s | 4.3 AB | $1.85 \mathrm{~s}$ |
| 34 | $\begin{gathered} 3.65 \mathrm{~d} \\ (4.8) \end{gathered}$ | $\begin{gathered} 4.2 \mathrm{dd} \\ (2.8,4.8) \end{gathered}$ | $\begin{array}{r} 5.3 \mathrm{~d} \\ (2.8) \end{array}$ | $1.9,2.3 \mathrm{ABX}$ | ${ }_{(5)}^{5.25 \mathrm{ABX}}$ | $\begin{aligned} & 5.75 \mathrm{~m} \\ & (5.5) \end{aligned}$ | $4.35 \mathrm{~d}$ <br> (5) | $\begin{aligned} & 2.8,3.05 \mathrm{AB} \\ & (3.8) \end{aligned}$ | 0.8 s | $\begin{aligned} & 4.1,4.3 \mathrm{AB} \\ & (12.2) \end{aligned}$ | $1.75 \mathrm{~s}$ |
| 35 | $\begin{gathered} 3.7 \mathrm{~d} \\ (4.7) \end{gathered}$ | $\begin{gathered} 4.2 \mathrm{dd} \\ (3,5) \end{gathered}$ | (3) | $\begin{aligned} & 2.45 \mathrm{ABX} \\ & (4.5,14.5) \end{aligned}$ | $\begin{aligned} & 5.15 \mathrm{ABX} \\ & (4.5) \end{aligned}$ | $\begin{aligned} & 5.9 \mathrm{~m} \\ & (5.5) \end{aligned}$ | $\begin{aligned} & 4.3 \mathrm{~d} \\ & (5.5) \end{aligned}$ | $\begin{aligned} & 2.85,3.1 \mathrm{AB} \\ & (3.9) \end{aligned}$ | 0.8 s | $\begin{aligned} & 4.1,4.4 \mathrm{AB} \\ & (12.3) \end{aligned}$ | $1.8 \mathrm{~s}$ |
| 36 | $\begin{gathered} 3.8 \mathrm{~d} \\ (4.8) \end{gathered}$ | $\begin{gathered} 4.25 \mathrm{dd} \\ (2.8,4.8) \end{gathered}$ | $\begin{aligned} & 5.15 \mathrm{~d} \\ & (2.8) \end{aligned}$ | $\begin{aligned} & 2.5,2.9 \mathrm{AB} \\ & (15.2) \end{aligned}$ | , | $\begin{gathered} 6.65 \mathrm{~m} \\ (1.4,5.6) \end{gathered}$ | $\begin{aligned} & 4.55 \mathrm{~d} \\ & (5.3) \end{aligned}$ | $\begin{aligned} & 2.85,3.1 \mathrm{AB} \\ & (3.7) \end{aligned}$ | 0.9 s | $\begin{aligned} & 4.2 \mathrm{AB} \\ & (12.2) \end{aligned}$ | $1.85 \mathrm{~s}$ |
| 37 | $\begin{gathered} 3.6 \mathrm{~d} \\ (4.8) \end{gathered}$ | 4.1 dd | $\begin{aligned} & 5.15 \mathrm{~d} \\ & (2.8) \end{aligned}$ | 2.0 ABX | 4.1 ABX | $\begin{aligned} & 5.5 \mathrm{~m} \\ & (5.1) \end{aligned}$ | 4.1 d | $\begin{aligned} & \text { 2.6, } 2.9 \mathrm{AB} \\ & (3.9) \end{aligned}$ | $0.75 \mathrm{~s}$ | $\begin{aligned} & 3.9,4.15 \mathrm{AB} \\ & (12.2) \end{aligned}$ | $1.7 \mathrm{~s}$ |
| 38 | $\begin{gathered} 3.7 \mathrm{~d} \\ (4.6) \end{gathered}$ | $\begin{gathered} 4.1 \mathrm{dd} \\ (2.4,4.6) \end{gathered}$ | $\begin{gathered} 5.25 \mathrm{~d} \\ (2.4) \end{gathered}$ | 2.0 ABX | $\begin{gathered} 5.0 \mathrm{~m} \\ (2.4) \end{gathered}$ | - | - | $\begin{aligned} & 2.8,3.1 \mathrm{AB} \\ & (3.8) \end{aligned}$ | $0.75 \mathrm{~s}$ | $\begin{aligned} & 4.0,4.5 \mathrm{AB} \\ & (12.0) \end{aligned}$ | $0.9 \mathrm{~s}$ |
| 39 | $\begin{aligned} & 3.65 \mathrm{~d} \\ & (4.6) \end{aligned}$ | $\begin{gathered} 4.1 \mathrm{dd} \\ (2.8,4.6) \end{gathered}$ | $\begin{aligned} & 5.35 \mathrm{~d} \\ & (2.8) \end{aligned}$ | $2.0 \mathrm{~m}$ | - | - | $4.1 \mathrm{~m}$ | $\begin{aligned} & 2.75,3.1 \mathrm{AB} \\ & (3.8) \end{aligned}$ | $0.75 \mathrm{~s}$ | $\begin{aligned} & \text { 4.2 AB } \\ & (13.3) \end{aligned}$ | $0.9 \mathrm{~s}$ |
| 40 | $\begin{aligned} & 3.75 \mathrm{~d} \\ & (4.5) \end{aligned}$ | $\begin{gathered} 4.15 \mathrm{dd} \\ (2.0,4.5) \end{gathered}$ | $\begin{aligned} & 5.25 \mathrm{~d} \\ & (2.0) \end{aligned}$ | 2.1 ABX | $\begin{aligned} & 5.25 \mathrm{ABX} \\ & (5.0) \end{aligned}$ | $\begin{gathered} 3.3 \mathrm{~d} \\ (5.5) \end{gathered}$ | $\begin{gathered} 4.3 \mathrm{~d} \\ (5.5) \end{gathered}$ | $\begin{aligned} & 2.7,3.05 \mathrm{AB} \\ & (4.0) \end{aligned}$ | $0.7 \mathrm{~s}$ | $\begin{aligned} & 3.3,4.0 \mathrm{AB} \\ & (12.5) \end{aligned}$ | $1.3 \mathrm{~s}$ |
| 42 | $\begin{aligned} & 3.9 \mathrm{~d} \\ & (5.0) \end{aligned}$ | $\begin{gathered} 5.1 \mathrm{dd} \\ (3.2,5.0) \end{gathered}$ | $\begin{aligned} & 5.8 \mathrm{~d} \\ & (3.5) \end{aligned}$ | $\begin{aligned} & \text { 2.4 ABX } \\ & (3.0,8.0) \end{aligned}$ | $\begin{aligned} & \text { 5.1 ABX } \\ & (3.0) \end{aligned}$ | $\begin{gathered} 5.1 \mathrm{~d} \\ (8.0) \end{gathered}$ | $\begin{gathered} 4.2 \mathrm{~d} \\ (8.0) \end{gathered}$ | $\begin{aligned} & 2.75,3.0 \mathrm{AB} \\ & (3.5) \end{aligned}$ | $0.6 \mathrm{~s}$ | $\begin{aligned} & 4.0 \mathrm{AB} \\ & (9.5) \end{aligned}$ | $1.15 \mathrm{~s}$ |
| 44 | $\begin{aligned} & 3.65 \mathrm{~d} \\ & (4.5) \end{aligned}$ | $\begin{gathered} 4.8 \mathrm{dd} \\ (2.8,5.5) \end{gathered}$ | $\begin{aligned} & 5.05 \mathrm{~d} \\ & (2.8) \end{aligned}$ | 2.1 ABX | 4.1 ABX | $\begin{gathered} 5.6 \mathrm{~m} \\ (5.5) \end{gathered}$ | 4.1 d | $2.8,3.05 \mathrm{AB}$ | $0.85 \mathrm{~s}$ | $\begin{aligned} & 3.5 \mathrm{AB} \\ & (12.4) \end{aligned}$ | $1.85 \mathrm{~s}$ |
| 45 | $\begin{aligned} & 3.55 \mathrm{~d} \\ & (5.0) \end{aligned}$ | $\begin{aligned} & 4.2 \mathrm{dd} \\ & (5.0) \end{aligned}$ | $4.2 \mathrm{~d}$ | $1.55,2.4 \mathrm{ABX}$ | $\begin{aligned} & \text { 4.2 ABX } \\ & (7.6) \end{aligned}$ | $4.2 \mathrm{~d}$ | $4.2 \mathrm{~d}$ | $\begin{aligned} & 2.75,2.9 \mathrm{AB} \\ & (4.2) \end{aligned}$ | $0.7 \mathrm{~s}$ | $\begin{aligned} & 3.7,3.9 \mathrm{AB} \\ & (8.9) \end{aligned}$ | $1.3 \mathrm{~s}$ |
| 46 | $\begin{gathered} 4.0 \mathrm{~d} \\ (4.8) \end{gathered}$ | $\begin{gathered} 5.3 \mathrm{dd} \\ (3.7,4.8) \end{gathered}$ | $\begin{gathered} 5.6 \mathrm{~d} \\ (3.7) \end{gathered}$ | $\begin{aligned} & 1.45,2.35 \mathrm{ABX} \\ & (8.4,14.5) \end{aligned}$ | $\begin{aligned} & \text { 5.4 ABX } \\ & (8.4) \end{aligned}$ | $\begin{aligned} & 4.35 \mathrm{~d} \\ & (8.2) \end{aligned}$ | $\begin{aligned} & 4.15 \mathrm{~d} \\ & (8.2) \end{aligned}$ | $\begin{aligned} & 2.75,3.1 \mathrm{AB} \\ & (3.8) \end{aligned}$ | $0.6 \mathrm{~s}$ | $\begin{aligned} & 3.8,4.0 \mathrm{AB} \\ & (9.2) \end{aligned}$ | $1.3 \mathrm{~s}$ |
| 47 | $\begin{aligned} & 3.8 \mathrm{~d} \\ & (4.8) \end{aligned}$ | $\begin{gathered} 5.2 \mathrm{dd} \\ (3.4,4.5) \end{gathered}$ | $\begin{aligned} & 6.05 \mathrm{~d} \\ & (3.4) \end{aligned}$ | $\begin{aligned} & 1.2,2.3 \mathrm{ABX} \\ & (5.6,14.5) \end{aligned}$ | $\begin{aligned} & \text { 5.3 ABX } \\ & (5.5) \end{aligned}$ | $\begin{gathered} 5.8 \mathrm{~m} \\ (1.0,6.0) \end{gathered}$ | $\begin{gathered} 4.3 \mathrm{~d} \\ (6.0) \end{gathered}$ | $2.3,3.05 \mathrm{AB}$ | $0.8 \mathrm{~s}$ | $\begin{aligned} & 3.65,4.0 \mathrm{AB} \\ & (15.0) \end{aligned}$ | $1.75 \mathrm{~s}$ |
| 48 | $\begin{aligned} & 3.55 \mathrm{~d} \\ & (4.6) \end{aligned}$ | $\begin{gathered} 4.25 \mathrm{dd} \\ (3.1,4.8) \end{gathered}$ | $\begin{aligned} & 5.45 \mathrm{~d} \\ & (3.0) \end{aligned}$ | $\begin{aligned} & 1.8,2.3 \mathrm{ABX} \\ & (5.5,14.7) \end{aligned}$ | $\begin{aligned} & 5.35 \mathrm{ABX} \\ & (5.5) \\ & \hline \end{aligned}$ | $\begin{gathered} 5.85 \mathrm{~m} \\ (1.0,5.6) \end{gathered}$ | $\begin{gathered} 4.3 \mathrm{~d} \\ (5.6) \end{gathered}$ | $\begin{aligned} & 2.8,3.05 \mathrm{AB} \\ & (3.9) \end{aligned}$ | $\begin{aligned} & 0.85 \mathrm{~s} \\ & - \\ & \hline \end{aligned}$ | $\begin{aligned} & 3.6,3.9 \mathrm{AB} \\ & (12.6) \end{aligned}$ | $1.75 \mathrm{~s}$ |

tions modified from those reported by Burmeister ${ }^{16}$ and Chi et al. ${ }^{17}$ A simplified isolation procedure was adopted, which allowed the facile and consistent isolation of gram quantities of metabolites from the cultures. Previously reported procedures have utilized blending ${ }^{16}$ or drying and grinding ${ }^{17}$ of the moldy solid substrate prior to extraction of the metabolites, both potentially hazardous operations that also release large amounts of extraneous material. Extraction was accomplished quite simply by gentle shaking of the broken up substrate with dichloromethane. Crystalline T-2 toxin and neosolaniol were then obtained after one or two chromatographic purifications of the extract. Isolated yields of T-2 toxin were $2.5-3.5 \mathrm{~g}$ per kg

[^2]of rice when cultured at $15^{\circ} \mathrm{C}$, together with $0.5-0.7 \mathrm{~g}$ per kg of neosolaniol. Yields of neosolaniol could be increased to 1 g per kg by raising the incubation temperature to 27 ${ }^{\circ} \mathrm{C}$ but at the expense of lower T-2 production.
Modification at C3. Acetylation ${ }^{18}$ and oxidation ${ }^{19}$ of the C3 hydroxyl group in T-2 to 4 and the labile ketone 5 , respectively, have been reported. Reduction of 5 with sodium borohydride gave T-2 toxin plus an unresolved mixture of the $3 \beta$ epimer of T-2 (6) and the transesterified compound 7. The common stereochemistry in 6 and 7 was confirmed by treatment with acetic anhydride-pyridine to give a single acetylation product (8). Hydrolysis gave

[^3]Table V. ${ }^{13} \mathrm{C}$ NMR Spectral Data ( 60 MHz ) for T-2 Derivatives, Chemical Shifts

| compd | position |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 | 11 | 12 | 13 | 14 | 15 | 16 |
| 8 | 80.9 | 73.7 | 73.3 | 49.2 | 42.6 | 28.1 | 67.8 | 136.9 | 123.1 | 67.8 | 64.4 | 46.2 | 6.6 | 64.4 | 20.5 |
| 10 | 79.1 | - | 77.2 | 48.7 | 43.4 | 29.6 | 67.6 | 139.8 | 120.7 | 66.5 | 64.1 | 47.2 | 6.3 | 64.7 | 20.7 |
| 11 | 78.1 | 78.7 | 84.2 | 48.6 | 42.9 | 27.2 | 68.3 | 136.0 | 123.7 | 67.2 | 64.3 | 47.1 | 6.9 | 64.3 | 20.2 |
| 12 | 79.0 | 78.0 | 77.0 | 48.7 | 43.2 | 27.6 | 68.3 | 136.4 | 123.6 | 67.5 | 64.4 | 47.2 | 6.8 | 64.4 | 20.4 |
| 13 | 78.8 | 36.5 | 74.7 | 48.3 | 42.1 | 27.8 | 68.0/66.0 ${ }^{\text {a }}$ | 136.5 | 123.3 | 68.0/66.0 | 65.2 | 47.8 | 6.6 | 64.4 | 20.3 |
| 14 | 78.3 | 39.6 | 74.0 | 48.8 | 41.4 | 27.5 | 68.0 | 136.5 | 123.4 | 65.7 | 65.2 | 47.4 | 7.1 | 64.4 | 20.2 |
| 15 | 78.4 | 39.4 | 74.1 | 49.0 | 43.5 | 28.0 | 67.1/66.5 | 139.6 | 121.6 | 67.1/66.5 | 65.3 | 47.4 | 7.0 | 61.9 | 20.4 |
| 16 | 99.4 | - | - | 48.4 | 41.9 | 28.9 | - | - | 121.4 | 67.4 | 55.1 | 48.4 | 8.5 | 63.6 | - |
| 20 | 83 | 78 | 77 | 48.6 | 42.4 | 27.4 | 67.8 | 136.4 | 123.3 | 67.3 | 63.9 | 46.7 | 6.7 | 64.4 | 20.0 |
| 21 | 75.5 | 73.2 | 209.1 | 53.8 | 45.5 | 25.9 | 67.2 | 136.6 | 122.7 | 67.6 | 60.8 | 48.3 | 6.6 | 62.9 | 20 |
| 24 | 77.7 | 71.0 | 39.0 | 45.1 | 41.7 | 27.2 | 68.2/67.5 | 135.9 | 124.1 | 68.3/67.5 | 64.5 | 48.6 | 12.1 | 64.5 | 20.3 |
| 25 | 79.7 | 31.6 | - | 49.4 | 43.6 | 25.7 | 68.4 | 136.2 | 124.2 | 66.0 | 64.7 | 45.0 | 12.0 | 66.0 | 20.4 |
| 26 | 81.0 | 146.5 | 126.2 | 49.9 | 42.5 | 27.1 | 66.9 | 135.7 | 124.7 | 69.2 | 70.2 | 51.2 | 9.0 | 65.0 | 21.1 |
| 28 | 79.1 | 212.2 | 46.0 | 48.7 | 44.3 | 27.7 | 69.4 | 139.7 | 121.9 | 66.7 | 64.2 | 47.9 | 12.1 | 62.0 | 20.5 |
| 29 | 79.3 | 211.7 | 48.1 | 42.5 | 45.6 | 27.0 | 68.3 | 136.9 | 123.6 | 68.3 | 64.1 | 48.8 | 11.6 | 64.3 | 20.6 |
| 30 | 77.6 | 79.2 | 83.3 | 48.8 | 43.4 | 30.3 | 67.2 | 139.6 | 120.9 | 66.5 | 64.5 | 46.8 | 6.3 | 64.5 | 20.4 |
| 34 | 78.2 | 78.7 | 84.3 | 48.3 | 42.9 | 27.7 | 68.1 | 136.2 | 123.6 | 67.2 | 64.3 | 47.1 | 6.9 | 64.5 | 21.0 |
| 35 | 78.1 | 78.7 | 84.3 | 48.7 | 42.8 | 26.6 | 69.2 | 136.2 | 123.9 | 67.5 | 64.2 | 47.0 | 7.1 | 64.2 | 20.4 |
| 36 | 83.5 | 78.9 | 78.1 | 47.5 | 48.9 | 38.1 | 196.5 | 138.8 | 136.8 | 68.2 | 64.4 | 46.6 | 6.2 | 64.4 | 15.4 |
| 37 | 84.3 | 79.1 | 78.1 | 48.8 | 45.9 | 31.5 | 68.2/67.8 | 142.6 | 120.8 | 68.2/67.7 | 64.2 | 47.0 | 6.8 | 63.7 | 18.7 |
| 38 | 84 | 79 | 78 | 49.2 | 43.8 | 28.6 | $72.0 / 69.6$ | 27.8 | 30.5 | 72.0/69.6 | 64.6 | 47.5 | 6.7 | 65.1 | 17.3 |
| 39 | 84 | 79 | 78 | 49.2 | 44.3 | 35.0 | 30.3 | 21.9 | 25.0 | 70.2 | 64.7 | 47.7 | 6.6 | 64.1 | 21 |
| 40 | 77.7 | 77.1 | 83.7 | 47.0 | 42.6 | 25.8 | 68.9 | 57.8 | 58.9 | 67.6 | 63.7 | 48.6 | 6.5 | 65.5 | 18.7 |
| 42 | 68 |  | 78 | 45.8 | 41.3 | 29.6 | 68 | - | - | 78 | 63.7 | 46.6 | 5.6 | 66 | 19.7 |
| 44 | 84 | 78 | 77 | 49.2 | 44.9 | 28.2 | 68.6/66.2 | - | - | 68.6/66.2 | 64.4 | 47.0 | 6.8 | 62.1 | 20.5 |
| 45 | 80.8 | 80.8 | 80.0 | 45.8 | 42.3 | 30.9 | 68.0/67.5 | 74.7 | 55.0 | 68.0/67.5 | 66.0 | 46.6 | 6.7 | 64.7 | 20.8 |
| 46 | 78.3 | 77.6 | 77.6 | 45.8 | 41.7 | 28.3 | 70.1/67.6 | 73.6 | 52.0 | 70.1/67 | 65.9 | 46.4 | 5.6 | 63.4 | 20.0 |
| 47 | 79.7 | 77.8 | 77 | 47.2 | 44.2 | 27.6 | 68.2/67.3 | 135.8 | 124.0 | 68.2/67.3 | 64.2 | 48.4 | 6.1 | 63.3 | 20.1 |
| 48 | 84.3 | 78.7 | 77.6 | 47.2 | 44.4 | 27.3 | 68.3/67.2 | 135.8 | 124.3 | 68.3/67.2 | 64.4 | 48.7 | 6.7 | 63.2 | 20.3 |

${ }^{a}$ Assignments uncertain.
the tetrahydroxy compound 9. Acetylation of neosolaniol with 1 equiv of acetic anhydride afforded selective acetylation at the C 3 hydroxyl to give 10, together with minor amounts of the C8 acetylated product 11 and the tetraacetate 12. Deoxygenation of T-2 toxin at C3 was achieved by using the reagent of Barton and $\mathrm{McCombie}{ }^{20,21}$ to give 13. Additional modification of 13 was obtained by partial hydrolysis with ammonium hydroxide to give 14 and 15. Oxidation of ketone 5 with $m$-chloroperoxybenzoic acid gave a single lactone (16) via a regiospecific Baeyer-Villiger rearrangement. The structural assignment was confirmed by NMR decoupling experiments, which indicated that the proton on the carbon attached to two oxygens coupled to H11. Thus decoupling of H 11 indicated H 2 to be at $\delta 5.0$. Selective decoupling at this frequency collapsed the lowfield ${ }^{13} \mathrm{C}$ signal at $\delta 99.5$.

Modification at C4. Hydrolysis of T-2 to a mixture of HT-2 (17), T-2 triol (18), and T-2-tetraol (19) (65:15:20) was achieved as reported previously. ${ }^{22}$ HT-2 was selectively acetylated to the isomer of T-2 (20), with only minor amounts of T-2 and the triacetate (4) being formed. Oxidation of 20 with chromium trioxide-pyridine afforded the stable crystalline ketone 21 . Reduction of 21 with sodium borohydride proceeded stereospecifically at the $\beta$-face of the molecule, but the product was not the expected carbinol 22 but the transesterified isomer 23. Support for the structural assignment was obtained by the failure of the product to oxidize back to 21 , unstable oxidation products being formed consistent with the C3 assignment. The isomer of T-2 (20) was deoxygenated under standard conditions to give 24.

[^4]Modification at C3 and C4. Deoxygenation of HT-2 (17) using the Barton procedure afforded a mixture of the simple C3-C4 dideoxygenated product 25 plus the diolefin 26, resulting from monodeoxygenation and elimination. Treatment of the dimesylate 27 with sodium methoxide gave the C3 deoxygenated ketone 28 in a reaction analogous to that reported for DAS by Sigg et al. ${ }^{23}$ Acetylation of 28 gave 29 .
Modification at C8. In order to prepare a series of C8-modified analogues of T-2, neosolaniol was used as starting material. Selective protection of the C 3 hydroxyl was achieved by silylation with tert-butyldimethylsilyl chloride and imidazole in dimethylformamide to give $30 .{ }^{24}$ Only minor amounts of the C8-silylated and C3,C8-disilylated products were formed. Acylation of 30 with a series of anhydrides or acid chlorides, followed by removal of the protective group with tetra- $n$-butylammonium fluoride, gave the C8-modified T-2 analogues 11 and 31-35. Oxidation of neosolaniol with pyridinium dichromate in dichloromethane afforded the enone 36 , which on reduction gave a mixture of neosolaniol and its $\mathrm{C} 8 \beta-\mathrm{OH}$ epimer 37.

Modification at C9-C10. Catalytic reduction of T-2 over $5 \%$ palladium on charcoal gave the expected reduction product $38^{18}$ plus the C 8 -deoxygenated product 39 . Although this reaction implies the intermediacy of DAS, attempts to modify the reaction conditions to achieve the direct conversion of T-2 to DAS were unsuccessful. An alternative procedure for achieving this conversion has been reported elsewhere. ${ }^{25}$ Oxidation of T-2 with mchloroperoxybenzoic acid gave the diepoxide 40 . Confirmation of the assignment of $\beta$ stereochemistry to the ep-

[^5]oxide was obtained by treatment of 40 with base to give the cyclized product 41, formed by intramolecular rearside attack at C 9 by the C 15 hydroxyl function. An analogous reaction was reported for a derivative of diacetoxyscirpenol. ${ }^{23}$ Acetylation of 41 gave the tetraacetate 42. Reaction of T-2 with a catalytic amount of osmium tetraoxide in aqueous acetone containing $N$-methylmorpholine N -oxide resulted in cis hydroxylation of the double bond to give 43.

Modification at C15. Treatment of neosolaniol with aqueous ammonia afforded regiospecific hydrolysis of the primary ester function at C15 to give triol 44. Acetylated analogues 47 and 48 were synthesized via the intermediacy of the bridged compound 45, following the methodology reported by Roush. ${ }^{26}$ Acylation of 45 gave 46, which on treatment with a zinc-silver couple gave two products identified as 47 and 48. Attempts to deoxygenate at C15, e.g., in compound 47 using the Barton method, were unsuccessful.

## Cytotoxicity Testing

Compounds were assayed for cytotoxicity to mouse lymphoma cells L5178Y as described in the Experimental Section. The results are shown in Tables I-III. Results are expressed in terms of $\mathrm{ng} / \mathrm{mL}$ since variations in molecular weight are generally small in relation to the accuracy of the assay.

## Results and Discussion

Modification at C3. The most cytotoxic of the natural metabolites amongst the simple trichothecenes are T-2 and DAS, both of which possess a $3 \alpha$-hydroxyl substituent. Acetylation of $\mathrm{C} 3(4,12,20)$ generally decreased activity by an order of magnitude, the exception being 3 -acetylneosolaniol (10), which did not differ significantly in activity from neosolaniol; deoxygenation at C3 decreased activity by 1 to 2 orders of magnitude ( $13,14,15$ ), the decrease being more pronounced with an adjacent $4 \beta$ hydroxyl group as in 14. Oxidation of the $3 \alpha$-hydroxyl to the ketone 5 or altering the stereochemistry to a $3 \beta$ configuration ( 8,9 ) resulted in similar reductions in activity. A free hydroxyl group in the natural $\alpha$ configuration therefore appears optimum for cytotoxicity in the simple nonmacrocyclic trichothecenes (it should be noted, however, that very potent macrocyclic trichothecenes such as verrucarin $\mathrm{A}^{27}$ are unsubstituted at C 3 ).

Modification at C4. T-2 and DAS possess a $4 \beta$-acetoxy substituent. Selective hydrolysis of the $4 \beta$-acetoxy function $(17,20)$ made little difference to activity in the presence of neighboring 3 -hydroxy or acetoxy groups, as observed previously for HT-2 (17) in other in vitro systems. ${ }^{6,11}$ A significant reduction in activity was observed when there was no adjacent substituent at C 3 as in 14. Deoxygenation at C 4 reduced activity by up to an order of magnitude (24), and oxidation (21) or alteration of the stereochemistry at C4 (23) led to substantial decreases in activity. The optimum substitution at C 4 therefore appears to be an acetoxy or hydroxyl function in the $\beta$ configuration. Simultaneous modification of substituents at C3 and C4 generally produced marked decreases in activity (16, 25 , 26, 28, 29).

Modification at C8. The fact that T-2 possesses a relatively bulky isovaleroyloxy substituent at C 8 , whereas diacetoxyscirpenol is unsubstituted, suggests that C8 corresponds to a region of steric tolerance in the interaction

[^6]of trichothecenes with their receptor. This was supported by the homologous series of C 8 esters 11 and $31-35$, which showed no significant differences in cytotoxicity from T-2 or DAS. Similarly the 8 -keto analogue 36 showed only a small reduction in activity as found in other in vitro test systems. ${ }^{11,12}$ The only compound modified at C8 that showed significantly reduced activity ( 20 fold) was the $8 \alpha$-hydroxy analogue neosolaniol (3). This may be due to reduced transport through the cell membrane rather than to stereoelectronic effects at the receptor. In contrast to alterations at C3 and C4, alteration of the stereochemistry as in the $8 \beta$-hydroxyl compound 37 made no significant difference to activity.

Modification at C9-C10. In line with Grove and Mortimer's observation ${ }^{5}$ with diacetoxyscirpenol, hydrogenation of the olefinic bond in T-2 led to a 10 -fold reduction in cytotoxicity (38). Epoxidation or hydroxylation of the double bond resulted in greater losses in activity (40, 43). Linkage of C15 to C9, which makes conformational as well as simple steric alterations to the molecule, substantially reduced or abolished activity (41, 42, 45, 46).

Modification at C15. T-2 and diacetoxyscirpenol each contain an acetoxy function at C 15 . Because of difficulties encountered in deoxygenation at C 15 , the only modification studied was that of hydrolysis to leave a free hydroxyl. This generally reduced activity ( $47,48,18$ ), except in the case of the neosolaniol analogue 44. Kaneko et al. ${ }^{12}$ reported that oxidation to give an aldehyde function at C15 increased protein synthesis inhibitory activity in HeLa cells.

## Conclusions

It appears that for the simple trichothecenes natural evolution has produced the metabolites T-2 toxin and diacetoxyscirpenol with maximum cytotoxicity. Modifications at C3, C4, C15, and C9-C10 generally decreased activity, marked dependence on stereochemistry being shown for substituents at C3 and C4. The only compounds synthesized that were equipotent with T-2 were a series of C8-modified ester analogues, suggesting that C 8 corresponds to a region of steric tolerance in the interaction of trichothecenes with a cellular receptor. However, it is not known to what extent mouse lymphoma cells metabolize trichothecenes and this could involve hydrolysis at C8. Lipophilicity is likely to effect the access of the compounds to the cell, and decreased activity was generally observed with compounds containing more than two free hydroxyl functions.

## Experimental Section

Melting points were determined in capillary tubes on a Gallenkamp apparatus and are uncorrected. ${ }^{1} \mathrm{H}$ NMR spectra were determined in deuteriochloroform unless otherwise stated with tetramethylsilane as internal standard on a JEOL JNM-MH-100 spectrometer. ${ }^{13} \mathrm{C}$ NMR spectra were obtained using a JEOL FX60Q spectrometer with deuteriochloroform as internal standard. IR spectra were measured in KBr disks on a PerkinElmer 157 spectrophotometer. Mass spectra were obtained on a VG 7070 EQ mass spectrometer. Optical rotations were measured in ethanol unless otherwise stated on a Bellingham and Stanley P-70 polarimeter. Column chromatography was performed under slight pressure (hand bellows) using Merck silica gel 60G ( $230-400$ mesh). Petroleum ether refers to the fraction with $\mathrm{bp} 60-80^{\circ} \mathrm{C}$. Trichothecenes may cause necrotic lesions of the skin at $\mu \mathrm{g}$ doses. All operations were performed in an efficiently ventilated fumecupboard using protective gloves. Glassware was decontaminated by standing in sodium hypochlorite solution overnight. Solid waste was sealed in polyethylene bags and incinerated.

Production of T-2 Toxin and Neosolaniol. Spores from a soil tube of $F$. sporotrichioides NRRL 3299 were plated onto Petri
dishes containing potato agar and grown at $25^{\circ} \mathrm{C}$ for 3 days. Subcultures were made onto potato agar slopes ( $7-10 \mathrm{~mL}$ ) contained in $1-\mathrm{oz}$ Mckartney bottles. After incubating for 1 week the slope cultures were transferred, one per flask, into $64 \times 250$ mL conical flasks each containing rice ( 50 g dry weight), previously soaked in an equal weight of distilled water for 30 min at room temperature, strained through a sieve, and autoclaved at 15 psi for 30 min . The flasks were incubated statically at $15^{\circ} \mathrm{C}$ for 4 weeks, after which time a mass of white mycelium with varying amounts of red and yellow pigmentation was visible. The matted rice and mycelium was broken up with a glass rod and extracted by gentle shaking ( 135 rpm ) overnight with dichloromethane ( 150 mL per flask), using a New Brunswick incubator-shaker. The extracts were decanted through Whatman 113v fluted filter paper, periodically changing the paper to ensure rapid filtration. The flasks were again extracted with a further portion of dichloromethane ( 80 mL ), and the contents were filtered through Celite contained in a Buchner funnel. The residue was washed several times with dichloromethane, and the combined filtrates were concentrated to an oil on a rotary evaporator. The residual oil was extracted with warm ether, monitoring by TLC to indicate complete extraction of the T-2 and neosolaniol. The extracts were dried $\left(\mathrm{MgSO}_{4}\right)$, concentrated, and purified by flash chromatography. The concentrate was dissolved in the minimum volume of dichloromethane and loaded onto a column ( $2 \mathrm{in} . \times 8 \mathrm{in}$.) containing silica gel ( 130 g ) and $1: 1$ ethyl acetate-petroleum ether as solvent. Elution with $1: 1$ ethyl acetate-ether yielded a crude fraction containing T-2 toxin, which crystallized on concentration and addition of petroleum ether. Recrystallization from etherpetroleum ether yielded T-2 toxin, $10.7 \mathrm{~g}, \mathrm{mp} 150-152^{\circ} \mathrm{C}$ (lit. ${ }^{18}$ $\operatorname{mp} 151-152^{\circ} \mathrm{C}$ ), pure by TLC and NMR: yield 3.35 g per kg of rice (dry wt).

Further elution of the column with 7:3 ethyl acetate-petroleum ether yielded a fraction containing neosolaniol. This was concentrated and rechromatographed on silica gel ( 40 g ) using 7:3 ethyl acetate-petroleum ether as eluant. Neosolaniol was crystallized from the appropriate fraction after concentration and addition of diisopropyl ether. Recrystallization from ethyl ace-tate-petroleum ether yielded neosolaniol, $2.5 \mathrm{~g}, \mathrm{mp} 169-170^{\circ} \mathrm{C}$ (lit. ${ }^{28} \mathrm{mp} 171-172^{\circ} \mathrm{C}$ ), pure by TLC and NMR: yield 0.66 g per kg of rice. Alteration of the culture conditions to $27^{\circ} \mathrm{C}$ for 3 weeks increased the isolated yield of neosolaniol to 1.04 g per kg of rice, but with decreased isolated yields of $\mathrm{T}-2(0.4-0.5 \mathrm{~g}$ per kg of rice).
$4 \beta, 15$-Diacetoxy- $3 \beta$-hydroxy- $8 \alpha$-[(3-methylbutyryl)oxy]-12,13-epoxytrichothec-9-ene (6). To a solution of $4 \beta, 15$-di-acetoxy- $8 \alpha$-[(3-methylbutyryl) oxy]-12,13-epoxytrichothec-9-en3 -one ( 5 ) ${ }^{19}$ ( $250 \mathrm{mg}, 0.54 \mathrm{mmol}$ ) in 2-propanol ( 30 mL ) was added sodium borohydride ( $4.9 \mathrm{mg}, 0.13 \mathrm{mmol}$ ) in 2-propanol ( 4.9 mL ), and the reaction was stirred at room temperature for 30 min . Acetone ( 5 mL ) was then added, and the mixture was concentrated and partitioned between diethyl ether and water. The combined ether extracts were dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated, and the residue was chromatographed (eluant $0.5 \%$ 2-propanol in benzene) to give 53 mg of 1 plus $47 \mathrm{mg}(19 \%)$ of an unresolved mixture of 6 and its transesterified isomer 7.
$3 \beta, 4 \beta, 15-T r i a c e t o x y-8 \alpha-[(3-m e t h y l b u t y r y l) o x y]-12,13-e p-$ oxytrichothec-9-ene (8). A mixture of 6 and $7(40 \mathrm{mg})$ was treated with acetic anhydride ( 1 mL ) and pyridine ( 1 mL ) at room temperature overnight. After concentration, the residue was chromatographed (eluant $4: 1$ petroleum ether-acetone) to give $33 \mathrm{mg}(82 \%)$ of $8:[\alpha]_{\mathrm{D}}+10.2^{\circ}$ (c 0.23 ): IR $1750 \mathrm{~cm}^{-1}$; MS, $m / z$ $526\left(\mathrm{MNH}_{4}{ }^{+}\right), 509\left(\mathrm{MH}^{+}\right)$. Anal. $\left(\mathrm{C}_{26} \mathrm{H}_{36} \mathrm{O}_{10}\right) \mathrm{C}, \mathrm{H}$.
$3 \beta, 4 \beta, 8 \alpha, 15-T e t r a h y d r o x y-12,13$-epoxytrichothec-9-ene (9). The reduction product from the ketone $5(400 \mathrm{mg})$ was stirred in methanol-water ( $4: 1,50 \mathrm{~mL}$ ) with potassium hydroxide ( 1 g ) at room temperature for 4 h . After concentration, the residue was chromatographed (initial eluant $5 \% \mathrm{MeOH}$ in $\mathrm{CHCl}_{3}$, final eluant $15 \% \mathrm{MeOH}$ in $\mathrm{CHCl}_{3}$ ) to give 75 mg of $9(29 \%)$ : $[\alpha]_{\mathrm{D}}$ $-58.6^{\circ}$ ( $c 0.13$ ); IR $3350 \mathrm{~cm}^{-1}$; MS, $m / z 316\left(\mathrm{MNH}_{4}{ }^{+}\right), 299\left(\mathrm{MH}^{+}\right)$.
$3 \alpha, 4 \beta, 15-$ Triacetoxy- $8 \alpha$-hydroxy-12,13-epoxytrichothec-9ene (10). Neosolaniol ( $3,220 \mathrm{mg}, 0.58 \mathrm{mmol}$ ) was treated with acetic anhydride ( $58 \mathrm{mg}, 0.57 \mathrm{mmol}$ ) and pyridine ( 1 mL ) at room

[^7]temperature overnight. The residue was chromatographed (eluant $1 \% \mathrm{MeOH}$ in $\mathrm{CHCl}_{3}$ ) to give $216 \mathrm{mg}(86 \%)$ of 10 as an oil: $[\alpha]_{\mathrm{D}}$ $+35.4^{\circ}$ ( $c 0.7$ ); IR $3540,1750 \mathrm{~cm}^{-1} ; \mathrm{MS}, m / z 442\left(\mathrm{MNH}_{4}{ }^{+}\right), 425$ ( $\mathrm{MH}^{+}$). Anal. $\left(\mathrm{C}_{21} \mathrm{H}_{28} \mathrm{O}_{9}\right) \mathrm{C}, \mathrm{H}$. The isomeric compound 11 and tetraacetate 12 were also obtained in $2 \%$ and $8 \%$ yield, respectively.

4 $\beta, 15$-Diacetoxy- $8 \alpha$-[(3-methylbutyryl)oxy]-12,13-epoxy-trichothec-9-ene (13). T-2 toxin ( $280 \mathrm{mg}, 0.6 \mathrm{mmol}$ ) and $N, N$ thiocarbonyldiimidazole ( $260 \mathrm{mg}, 1.44 \mathrm{mmol}$ ) in dichloromethane ( 5 mL ) were heated under reflux for 1.5 h . After cooling, the mixture was concentrated and chromatographed (eluant $\mathrm{CHCl}_{3}$ ) to give $4 \beta, 15$-diacetoxy-3-[(1-imidazolylthiocarbonyl)oxy]-8 $\alpha$ -[(3-methylbutyryl)oxy]-12,13-epoxytrichothec-9-ene ( 335 mg , $97 \%$ ); IR $1750 \mathrm{~cm}^{-1}$. A solution of this compound ( $270 \mathrm{mg}, 0.48$ mmol ) in anhydrous toluene ( 18 mL ) was added dropwise over a period of 90 min to a refluxing solution of tri- $n$-butyltin hydride ( $390 \mathrm{mg}, 1.34 \mathrm{mmol}$ ) in anhydrous toluene ( 12 mL ) containing a catalytic amount ( 0.04 mmol ) of $\alpha$-azoisobutyrylnitrile. After refluxing an additional hour, the reaction was cooled, concentrated, and chromatographed (initial eluant petroleum ether, final eluant $20 \%$ acetone in petroleum ether) to give 270 mg of $13(62 \%): \mathrm{mp}$ $105-6^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}-19.7^{\circ}(c 0.6)$; IR $1750 \mathrm{~cm}^{-1}, \mathrm{MS}, m / z 468\left(\mathrm{MNH}_{4}{ }^{+}\right)$, $451\left(\mathrm{MH}^{+}\right)$. Anal. $\left(\mathrm{C}_{24} \mathrm{H}_{34} \mathrm{O}_{8} \cdot 0.25 \mathrm{H}_{2} \mathrm{O}\right) \mathrm{C}, \mathrm{H}$.

15-Acetoxy-4 $\beta$-hydroxy- $8 \alpha$-[(3-methylbutyryl)oxy]-12,13-epoxytrichothec-9-ene (14). $13(90 \mathrm{mg}, 0.2 \mathrm{mmol}$ ) was added to a solution of $2 \%$ ammonia in methanol-water ( $4: 1,45 \mathrm{~mL}$ ), and the solution was stirred at room temperature for 48 h . Concentration and chromatography (initial eluant $\mathrm{CHCl}_{3}$, final eluant $5 \% \mathrm{MeOH}$ in $\mathrm{CHCl}_{3}$ ) gave 10 mg of $14(16 \%):[\alpha]_{\mathrm{D}}-31.6^{\circ}$ (c 0.1): IR $3500,1740 \mathrm{~cm}^{-1}$; MS, $m / z 426\left(\mathrm{MNH}_{4}^{+}\right), 409\left(\mathrm{MH}^{+}\right)$. Further elution yielded the triol 15 ( $11 \mathrm{mg}, 25 \%$ ): mp 171-172 ${ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}-76.0^{\circ}$ (c 0.5); IR $3500 \mathrm{~cm}^{-1} ; \mathrm{MS}, m / z 300\left(\mathrm{MNH}_{4}{ }^{+}\right), 283$ ( $\mathrm{MH}^{+}$). Anal. $\left(\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{O}_{5}\right) \mathrm{C}, \mathrm{H}$.

Lactone 16. $m$-Chloroperoxybenzoic acid ( $35 \mathrm{mg}, 0.2 \mathrm{mmol}$ ) was added to a solution of the ketone $5(90 \mathrm{mg}, 0.19 \mathrm{mmol})$ in dichloromethane ( 7 mL ), and the mixture was stirred at room temperature for 5 h . The solution was washed with $20 \%$ sodium thiosulfate solution and then with 2 N sodium bicarbonate. After drying $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and removal of the solvent, the crude product was chromatographed (eluant $1 \% \mathrm{MeOH}$ in $\mathrm{CHCl}_{3}$ ) to give 67 mg of 16 ( $72 \%$ ): $[\alpha]_{\mathrm{D}}-19.6^{\circ}$ (c 0.7); IR $1750 \mathrm{~cm}^{-1} ; \mathrm{MS}, m / z$ 498.2337 (498.2339) ( $\mathrm{MNH}_{4}^{+}$), $481\left(\mathrm{MH}^{+}\right)$
$3 \alpha$,15-Diacetoxy- $4 \beta$-hydroxy- $8 \alpha$-[(3-methylbutyryl)oxy]-12,13-epoxytrichothec-9-ene (20) was prepared from HT-2 (17) ${ }^{22}$ and 1 equiv of $\mathrm{Ac}_{2} \mathrm{O}$ in pyridine: yield $64 \% ; \mathrm{mp} 147.5-148.5 ;[\alpha]_{\mathrm{D}}$ $-24.9^{\circ}$ (c 0.5); IR $3490,1750 \mathrm{~cm}^{-1}$; MS, $m / z 484\left(\mathrm{MNH}_{4}{ }^{+}\right)$. Anal. $\left(\mathrm{C}_{24} \mathrm{H}_{34} \mathrm{O}_{9}\right) \mathrm{C}, \mathrm{H}$.

3 $\alpha$,15-Diacetoxy-8 $\alpha$-[(3-methylbutyryl)oxy]-12,13-epoxy trichothec-9-en-4-one (21). Pyridine ( 1.4 mL ) was added dropwise to a suspension of chromium trioxide ( $800 \mathrm{mg}, 8 \mathrm{mmol}$ ) in dichloromethane ( 8 mL ) at $0^{\circ} \mathrm{C}$, and the mixture was stirred at this temperature for a further 20 min . A solution of 20 (110 $\mathrm{mg}, 0.24 \mathrm{mmol}$ ) in dichloromethane ( 2 mL ) was added, and the reaction was allowed to warm to room temperature. After 1 h TLC indicated complete absence of starting material. Diethyl ether was added and the resulting suspension was filtered through Celite. The filtrate was concentrated and the residue was crystallized from ether to give 90 mg of 21 : yield $82 \%$; mp 148-149 ${ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}+43.2^{\circ}(c 0.5)$; IR $1750 \mathrm{~cm}^{-1} ;$ MS, $m / z 482\left(\mathrm{MNH}_{4}{ }^{+}\right), 465$ $\left(\mathrm{MH}^{+}\right)$. Anal. $\left(\mathrm{C}_{24} \mathrm{H}_{32} \mathrm{O}_{9} \cdot 0.25 \mathrm{H}_{2} \mathrm{O}\right) \mathrm{C}, \mathrm{H}$.

4 $\alpha$,15-Diacetoxy- $3 \alpha$-hydroxy-8 $\alpha$-[(3-methylbutyryl)oxy]-12,13-epoxytrichothec-9-ene (23) was prepared from 21 by reduction with sodium borohydride in 2-propanol: yield $31 \%$; IR $3490,1750 \mathrm{~cm}^{-1} ; \mathrm{MS}, m / z 484\left(\mathrm{MNH}_{4}^{+}\right), 467\left(\mathrm{MH}^{+}\right)$.

3 $\alpha, 15$-Diacetoxy- $8 \alpha$-[(3-methylbutyryl)oxy]-12,13-ерoxy-trichothec-9-ene (24) was prepared from 20 by Barton deoxygenation: yield $73 \%$; $\operatorname{mp} 122.5-123.5^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}+2.5^{\circ}$ (c 0.5); IR $1750 \mathrm{~cm}^{-1} ; \mathrm{MS}, \mathrm{m} / \boldsymbol{z} 468\left(\mathrm{MNH}_{4}^{+}\right), 451\left(\mathrm{MH}^{+}\right)$. Anal. $\left(\mathrm{C}_{24} \mathrm{H}_{34} \mathrm{O}_{8}\right)$ $\mathrm{C}, \mathrm{H}$.

15-Acetoxy-8 $\alpha \cdot[(3-$ methylbutyryl)oxy]-12,13-epoxy-trichothec-9-ene (25) was prepared from HT-2 (17) by Barton deoxygenation and isolated by chromatography (eluant 8:2 petroleum ether-acetone): yield $31 \%$; $[\alpha]_{\mathrm{D}}-58.3^{\circ}$ (c 0.12); IR 1740 $\mathrm{cm}^{-1} ; \mathrm{MS}, m / z 410.2520(410.2543)\left(\mathrm{MNH}_{4}{ }^{+}\right), 393\left(\mathrm{MH}^{+}\right)$. The diene 26 was also obtained in $10 \%$ yield: $[\alpha]_{D}+129.8^{\circ}$ (c 0.6); IR $1730 \mathrm{~cm}^{-1}$; MS, $m / z 408.2432$ (408.2386) $\left(\mathrm{MNH}_{4}^{+}\right), 391\left(\mathrm{MN}^{+}\right)$.

15-Acetoxy- $3 \alpha, 4 \beta$-bis(mesyloxy)- $8 \alpha$-[(3-methylbutyryl)-oxy]-12,13-epoxytrichothec-9-ene (27). To a solution of HT-2 (17) $(50 \mathrm{mg}, 0.12 \mathrm{mmol})$ in anhydrous pyridine at $0^{\circ} \mathrm{C}$ was added methanesulfonyl chloride ( 0.5 mL ), and the mixture was stirred at this temperature for 2 h . Concentration and chromatography (eluant $\mathrm{CHCl}_{3}$ ) yielded 60 mg of 27 ( $88 \%$ ): IR $1730 \mathrm{~cm}^{-1}$; MS, $m / z 598\left(\mathrm{MNH}_{4}{ }^{+}\right)$.
$8 \alpha, 15$-Dihydroxy-12,13-e poxytrichothec-9-en-3-one (28). To a solution of $60 \%$ sodium hydride ( $40 \mathrm{mg}, 1 \mathrm{mmol}$ ) in methanol ( 3 mL ) was added the dimesylate $27(55 \mathrm{mg}, 0.09 \mathrm{mmol}$ ) in methanol ( 3 mL ), and the mixture was refluxed under argon for 3 h . After cooling, the reaction was quenched with a few drops of water and concentrated and the residue was chromatographed (initial eluant $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, final eluant $21 \% \mathrm{MeOH}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to give the ketone $28,20 \mathrm{mg}(75 \%)$ : $\mathrm{mp} 160.5-161.5^{\circ} \mathrm{C} ;[\alpha]_{D}-35.2^{\circ}$ (c $0.5)$; IR $3450,1750 \mathrm{~cm}^{-1}$; MS, $m / z 298\left(\mathrm{MNH}_{4}{ }^{+}\right), 281\left(\mathrm{MH}^{+}\right)$. Anal. $\left(\mathrm{C}_{15} \mathrm{H}_{20} \mathrm{O}_{5}\right) \mathrm{C}, \mathrm{H}$.
$8 \alpha, 15$-Diacetoxy-12,13-epoxytrichothec-9-en-3-one (29) was prepared from 28 by acetylation with excess $\mathrm{Ac}_{2} \mathrm{O}$ : yield $76 \%$; $[\alpha]_{\mathrm{D}}-54.9^{\circ}$ ( $c 0.5$ ); IR $3000,1740 \mathrm{~cm}^{-1}$; MS, $m / z 382\left(\mathrm{MNH}_{4}{ }^{+}\right)$, 365 ( $\mathrm{MH}^{+}$).
$3 \alpha$-(tert-Butyldimethylsiloxy)-4 $\beta, 8 \alpha, 15$-triacetoxy-12,13-epoxytrichothec-9-ene (30). A mixture of neosolaniol (3) (250 $\mathrm{mg}, 0.66 \mathrm{mmol}$ ), tert-butyldimethylsilyl chloride ( $300 \mathrm{mg}, 1.99$ mmol ), and imidazole ( $135 \mathrm{mg}, 1.99 \mathrm{mmol}$ ) in anhydrous dimethylformamide ( 8 mL ) was stirred at room temperature, and the reaction was monitored by TLC. After 3 days the solvent was removed and the residue was chromatographed (initial eluant $\mathrm{CHCl}_{3}$, final eluant $5 \% \mathrm{MeOH}$ in $\mathrm{CHCl}_{3}$ ) to give the title compound, $242 \mathrm{mg}(74 \%): \operatorname{mp} 157.5-158.5^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}+20.5^{\circ}(c 0.6)$; IR $3500,1735,1250 \mathrm{~cm}^{-1} ; \mathrm{MS}, m / z 514\left(\mathrm{MNH}_{4}{ }^{+}\right), 497\left(\mathrm{MH}^{+}\right)$. Anal. $\left(\mathrm{C}_{25} \mathrm{H}_{40} \mathrm{O}_{8} \mathrm{Si}\right) \mathrm{C}, \mathrm{H}$.

Also obtained were $8 \alpha$-(tert-butyldimethylsiloxy)- $4 \beta, 15$-di-acetoxy-3 $\alpha$-hydroxy-12,13-epoxytrichothec-9-ene ( $23 \mathrm{mg}, 7 \%$ ) [mp $173-174{ }^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}-16.3^{\circ}$ (c 0.5); IR $3480,1735,1250 \mathrm{~cm}^{-1}$; MS, $m / z 514\left(\mathrm{MNH}_{4}{ }^{+}\right), 497\left(\mathrm{MH}^{+}\right)$. Anal. $\left.\left(\mathrm{C}_{25} \mathrm{H}_{40} \mathrm{O}_{8} \mathrm{Si}\right) \mathrm{C}, \mathrm{H}.\right]$ and $3 \alpha, 8 \alpha$-bis (tert-butylsiloxy)-4 $\beta$,15-diacetoxy-12,13-epoxy-trichothec-9-ene ( $60 \mathrm{mg}, 15 \%$ ) [mp $116-118^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}+12.6^{\circ}(\mathrm{c}$ $0.6)$; IR $3490,1745,1250 \mathrm{~cm}^{-1}$; MS, $m / z 628\left(\mathrm{MNH}_{4}^{+}\right), 611\left(\mathrm{MH}^{+}\right)$. Anal. $\left.\left(\mathrm{C}_{31} \mathrm{H}_{54} \mathrm{O}_{8} \mathrm{Si}_{2}\right) \mathrm{C}, \mathrm{H}.\right]$.
$3 \alpha$-Hydroxy-4 $\beta, 8 \alpha, 15$-triacetoxy-12,13-epoxytrichothec-9ene (11). To a solution of $30(150 \mathrm{mg}, 0.3 \mathrm{mmol})$ in anhydrous dichloromethane ( 3 mL ) were added acetic anhydride ( $61 \mathrm{mg}, 0.60$ mmol ), triethylamine ( $61 \mathrm{mg}, 0.60 \mathrm{mmol}$ ), and a catalytic amount of 4 -(dimethylamino) pyridine ( $5 \mathrm{mg}, 0.04 \mathrm{mmol}$ ). The reaction was stirred overnight at room temperature, concentrated and chromatographed (eluant $10 \%$ acetone in petroleum ether) to give $3 \alpha$-(tert-butyldimethylsiloxy)-4 $\beta, 8 \alpha, 15$-triacetoxy-12,13-epoxy-trichothec-9-ene ( $134 \mathrm{mg}, 82 \%$ ); IR $1735,1250 \mathrm{~cm}^{-1}$. A solution of this compound ( $134 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) was stirred with tetra- $n$ butylammonium fluoride ( 0.25 mL of a 1 M solution in THF, 0.25 mmol ) in anhydrous tetrahydrofuran ( 2 mL ) at room temperature for 1 h . The solvent was removed, and the residue was taken up in ether, washed with water, and dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$. Concentration and chromatography (initial eluant $\mathrm{CHCl}_{3}$, final eluant $1 \% \mathrm{MeOH}$ in $\mathrm{CHCl}_{3}$ ) gave 94 mg of $11(89 \%)$ : $\mathrm{mp} \mathrm{182-183}{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}-3.6^{\circ}$ (c 0.5); IR $3350,1730,1250,1225 \mathrm{~cm}^{-1}$; MS, $m / z 442\left(\mathrm{MNH}_{4}{ }^{+}\right)$, $425\left(\mathrm{MH}^{+}\right)$. Anal. $\left(\mathrm{C}_{21} \mathrm{H}_{28} \mathrm{O}_{9}\right) \mathrm{C}, \mathrm{H}$.

The following $8 \alpha$-acylated trichothecenes were prepared similarly from 30.
$3 \alpha$-Hydroxy-4 $\beta, 15$-diacetoxy- $8 \alpha$-(propionyloxy)-12,13-ep-oxytrichothec-9-ene (31): overall yield $56 \%$; mp $179.5-180^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}+14.1^{\circ}(c 0.4)$; IR $3400,1740,1250 \mathrm{~cm}^{-1} ;$ MS, $m / z 456$ $\left(\mathrm{MNH}_{4}{ }^{+}\right), 439\left(\mathrm{MH}^{+}\right)$. Anal. $\left(\mathrm{C}_{22} \mathrm{H}_{30} \mathrm{O}_{9}\right) \mathrm{C}, \mathrm{H}$.
$3 \alpha$-Hydroxy-4 $\beta$,15-diacetoxy- $8 \alpha$-(butyryloxy)-12,13-epoxy-trichothec-9-ene (32): overall yield $64 \% ; \mathrm{mp} 148-149^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}$ $+12.9^{\circ}(c .0 .45)$; IR $3400,1730,1240 \mathrm{~cm}^{-1} ; \mathrm{MS}, m / z 470\left(\mathrm{MNH}_{4}^{+}\right)$, $453\left(\mathrm{MH}^{+}\right)$. Anal. $\left(\mathrm{C}_{23} \mathrm{H}_{32} \mathrm{O}_{9}\right) \mathrm{C}, \mathrm{H}$.
$3 \alpha$-Hydroxy- $4 \beta, 15$-diacetoxy- $8 \alpha$-(pentanoyloxy)-12,13-ep-oxytrichothec-9-ene (33): overall yield $58 \%$; $\mathrm{mp} 152.5-153{ }^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}+13.3^{\circ}$ (c 0.55 ); IR $3460,1740,1250 \mathrm{~cm}^{-1} ;$ MS, $m / z 484$ $\left(\mathrm{MNH}_{4}{ }^{+}\right), 467\left(\mathrm{MH}^{+}\right)$. Anal. $\left(\mathrm{C}_{24} \mathrm{H}_{34} \mathrm{O}_{9} \cdot 0.25 \mathrm{H}_{2} \mathrm{O}\right) \mathrm{C}$, H .
$3 \alpha$-Hydroxy- $4 \beta, 15$-diacetoxy- $\alpha$-(hexanoyloxy)-12,13-ep-oxytrichothec-9-ene (34): overall yield $55 \%$; mp $146-147^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}+13.8^{\circ}$ (c 0.5); IR 3450, 1735, $1250 \mathrm{~cm}^{-1}$; MS, $m / z 498$ $\left(\mathrm{MNH}_{4}{ }^{+}\right), 481\left(\mathrm{MH}^{+}\right)$. Anal. $\left(\mathrm{C}_{25} \mathrm{H}_{36} \mathrm{O}_{9} \cdot 0.25 \mathrm{H}_{2} \mathrm{O}\right) \mathrm{C}, \mathrm{H}$.
$3 \alpha$-Hydroxy- $4 \beta, 15$-diacetoxy- $8 \alpha$-(benzoyloxy)-12,13-ep-oxytrichothec-9-ene (35). To a solution of 30 ( $102 \mathrm{mg}, 0.21$ mmol ) in anhydrous acetonitrile ( 3 mL ) were added benzoyl chloride ( $89 \mathrm{mg}, 0.63 \mathrm{mmol}$ ), triethylamine ( $81 \mathrm{mg}, 0.8 \mathrm{mmol}$ ), and a catalytic amount of 4 -(dimethylamino) pyridine ( $5 \mathrm{mg}, 0.04$ mmol ). The reaction was heated under reflux for 7 h and concentrated, and the residue was chromatographed (eluant $2 \%$ MeOH in $\mathrm{CHCl}_{3}$ ) to give $8 \alpha$-(benzoyloxy)- $3 \alpha$-(tert-butyldi-methylsiloxy)-4 $\beta, 15$-diacetoxy-12,13-epoxytrichothec-9-ene (111 $\mathrm{mg}, 90 \%$ ); IR $1740,1720,1270,1250 \mathrm{~cm}^{-1}$. Desilylation was carried out as described above to give $35(68 \%)$ : mp $204-205^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}$ $-71.1^{\circ}$ (c 0.5); IR 3450, $1740,1720,1250 \mathrm{~cm}^{-1}$; MS, $m / z 504$ $\left(\mathrm{MNH}_{4}{ }^{+}\right), 487\left(\mathrm{MH}^{+}\right)$. Anal. $\left(\mathrm{C}_{26} \mathrm{H}_{30} \mathrm{O}_{9} \cdot 0.5 \mathrm{H}_{2} \mathrm{O}\right) \mathrm{C}, \mathrm{H}$.
4 $\beta, 15$-Diacetoxy- $3 \alpha$-hydroxy-12,13-epoxytrichothec-9-en8 -one (36) was prepared from neosolaniol and pyridinium dichromate: yield $90 \%$; $[\alpha]_{\mathrm{D}}+81.2^{\circ}$ (c 0.40 ); IR $3510,1755 \mathrm{~cm}^{-1}$; MS, $m / z 398\left(\mathrm{MNH}_{4}{ }^{+}\right), 381\left(\mathrm{MH}^{+}\right)$.

4 $\beta$,15-Diacetoxy- $3 \alpha, 8 \beta$-dihydroxy-12,13-epoxytrichothec9 -ene (37) was prepared by reduction of 36 with $\mathrm{NaBH}_{4}$ in 2propanol: yield $13 \%$; $[\alpha]_{D}+72.5^{\circ}$ (c 0.13 ); IR $3400,1740 \mathrm{~cm}^{-1}$; MS, $m / z 400.1944$ (400.1971) $\left(\mathrm{MNH}_{4}{ }^{+}\right)$.
4 $\beta, 15$-Diacetoxy- $3 \alpha$-hydroxy- $8 \alpha$-[(3-methylbutyryl)oxy]-12,13-epoxytrichothecane (38). A solution of T-2 $90 \mathrm{mg}, 0.19$ mmol ) in methanol ( 2 mL ) was added to a suspension of $10 \%$ palladium on charcoal ( 10 mg ), and the mixture was stirred at room temperature under an atmosphere of hydrogen for 5 h . After filtration and concentration the residue was chromatographed (initial eluant $\mathrm{CHCl}_{3}$, final eluant $5 \% \mathrm{MeOH}$ in $\mathrm{CHCl}_{3}$ ) to yield $37 \mathrm{mg}(41 \%)$ of 38 as a mixture of isomers: $\mathrm{mp} 132-4{ }^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}$ $+56.9^{\circ}$ ( 0.5 ); IR $3500,1725 \mathrm{~cm}^{-1}$; MS, $m / z 486\left(\mathrm{MNH}_{4}{ }^{+}\right)$. Anal. $\left(\mathrm{C}_{24} \mathrm{H}_{36} \mathrm{O}_{9}\right) \mathrm{C}, \mathrm{H}$.
Further elution yielded the C8-deoxygenated hydrogenolysis product 39 ( $35 \mathrm{mg}, 49 \%$ ): mp 137-8 ${ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}+44.5^{\circ}(c 0.5) \mathrm{IR}$ $1748 \mathrm{~cm}^{-1}$; MS, $m / z 386\left(\mathrm{MNH}_{4}{ }^{+}\right), 369\left(\mathrm{MH}^{+}\right)$. Anal. $\left(\mathrm{C}_{19} \mathrm{H}_{28}{ }^{-}\right.$ $\left.\mathrm{O}_{7} \cdot 0.25 \mathrm{H}_{2} \mathrm{O}\right) \mathrm{C}, \mathrm{H}$.
4 $\beta$,15-Diacetoxy- $3 \alpha$-hydroxy- $8 \alpha$-[(3-methylbutyryl)oxy]$9 \beta, 10 \beta: 12,13$-diepoxytrichothecane (40). To a solution of T-2 ( $130 \mathrm{mg}, 0.28 \mathrm{mmol}$ ) in ethanol-free chloroform ( 10 mL ) was added $m$-chloroperoxybenzoic acid ( $160 \mathrm{mg}, 0.93 \mathrm{mmol}$ ), and the mixture was heated under reflux for 48 h . After cooling, the solution was washed sequentially with $20 \%$ aqueous sodium thiosulfate and 2 N aqueous sodium bicarbonate. After drying $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentration, chromatography (initial eluant $\mathrm{CHCl}_{3}$, final eluant $2 \% \mathrm{MeOH}$ in $\mathrm{CHCl}_{3}$ ) yielded 100 mg of $40(76 \%): \mathrm{mp} 161-2{ }^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}+62.4^{\circ}$ (c 0.6 ); IR $1730 \mathrm{~cm}^{-1}$; MS, $m / z 500\left(\mathrm{MNH}_{4}{ }^{+}\right)$. Anal. $\left(\mathrm{C}_{24} \mathrm{H}_{34} \mathrm{O}_{10}\right) \mathrm{C}, \mathrm{H}$.
$3 \alpha, 4 \beta, 8 \alpha, 10 \beta$-Tetrahydroxy- $9 \alpha, 15: 12,13$-diepoxytrichothecane (41) was prepared from 40 by hydrolysis with KOH in aqueous methanol as used for the preparation of 9 : yield $56 \% ;[\alpha]_{\mathrm{D}}+9.3^{\circ}(c 0.4)$; MS, $m / z 332\left(\mathrm{MNH}_{4}{ }^{+}\right), 315\left(\mathrm{MH}^{+}\right)$. Acetylation gave $3 \alpha, 4 \beta, 8 \alpha, 10 \beta$-tetraacetoxy- $9 \alpha, 15: 12,13$-diepoxytrichothecane (42): yield $56 \%$; mp $159-160^{\circ} \mathrm{C} ;[\alpha]_{D}+73.7^{\circ}(c$ $0.55)$; IR $1740 \mathrm{~cm}^{-1}$; MS, $m / z 500\left(\mathrm{MNH}_{4}^{+}\right)$. Anal. $\left(\mathrm{C}_{23} \mathrm{H}_{30} \mathrm{O}_{7}\right)$ C, H.
$4 \beta, 15$-Diacetoxy- $8 \alpha$-[(3-methylbutyryl)oxy]-3 $\alpha, 9 \beta, 10 \beta$-tri-hydroxy-12,13-epoxytrichothecane (43). T-2 toxin ( $50 \mathrm{mg}, 0.11$ mmol ) was dissolved in $50 \%$ aqueous acetone ( 3 mL ) and a catalytic amount (one crystal) of osmium tetraoxide was added followed by $N$-methylmorpholine $N$-oxide ( $20 \mathrm{mg}, 0.17 \mathrm{mmol}$ ). The reaction was stirred at room temperature and monitored by TLC. Two further similar amounts of reagents were added at intervals of 6 h . After column chromatography (elution with $5 \%$ MeOH in $\mathrm{CHCl}_{3}$ ) the triol 43 was obtained ( $26 \mathrm{mg}, 48 \%$ ): $[\alpha]_{\mathrm{D}}$ $+62.4^{\circ}$ ( $\mathbf{c} 0.13$ ); IR $3480,1745 \mathrm{~cm}^{-1}$; MS, $m / z 518.2601$ (518.2713) $\left(\mathrm{MNH}_{4}{ }^{+}\right), 501\left(\mathrm{MH}^{+}\right)$.
$4 \beta$-Acetoxy- $3 \alpha, 8 \alpha, 15$-trihydroxy-12,13-epoxytrichothec-9ene (44) was prepared by treatment of neosolaniol with aqueous ammonia: yield $67 \%$; mp $163-4^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}-2.1^{\circ}$ (c 0.5); IR 3400, $1745 \mathrm{~cm}^{-1}$; MS, $m / z 358\left(\mathrm{MNH}_{4}{ }^{+}\right), 341\left(\mathrm{MH}^{+}\right)$. Anal. $\left(\mathrm{C}_{17} \mathrm{H}_{24} \mathrm{O}_{7}\right)$ $\mathrm{C}, \mathrm{H}$.
$10 \beta$-Bromo- $3 \alpha, 4 \beta, 8 \alpha$-trihydroxy- $9 \alpha, 15: 12,13$-diepoxytrichothecane (45). To a solution of T-2 tetraol (19) ( 90 mg , 0.30 mmol ) in anhydrous acetonitrile ( 87 mL ) was added $N$ bromosuccinimide ( $59 \mathrm{mg}, 0.33 \mathrm{mmol}$ ), and the mixture was stirred at room temperature overnight. Chromatography (initial eluant $\mathrm{CHCl}_{3}$, final eluant $5 \% \mathrm{MeOH}$ in $\mathrm{CHCl}_{3}$ ) yielded the pure bromide

45 ( $111 \mathrm{mg}, 97 \%$ ): IR $3450 \mathrm{~cm}^{-1} ; \mathrm{MS}, m / z 396,394\left(\mathrm{MNH}_{4}{ }^{+}\right)$, 378, $376\left(\mathrm{MH}^{+}\right)$.
$10 \beta$-Bromo- $3 \alpha, 4 \beta, 8 \alpha$-triace toxy- $9 \alpha, 15: 12,13$-diepoxytrichothecane (46) was prepared from 45 and excess $\mathrm{Ac}_{2} \mathrm{O}$ : yield $84 \% ;[\alpha]_{D}+27.1^{\circ}$ (c 0.56 ); IR $1745 \mathrm{~cm}^{-1} ;$ MS, $m / z 522,520.1240$ (520.1182) $\left(\mathrm{MNH}_{4}{ }^{+}\right), 503,501\left(\mathrm{MH}^{+}\right)$.
$3 \alpha, 4 \beta, 8 \alpha$-Triacetoxy-15-hydroxy-12,13-epoxytrichothecene (47). The zinc-silver couple was prepared immediately before use as described previously. ${ }^{26}$ To a suspension of the couple (prepared from zinc dust ( 30 g ) and silver acetate ( 175 mg ) ) in ether ( 73 mL ) was added 46 ( $942 \mathrm{mg}, 1.87 \mathrm{mmol}$ ) in a tetra-hydrofuran-ethanol mixture ( $175 \mathrm{~mL}, 5: 1$ ). The mixture was heated under reflux for 16 h . The solvents were removed in vacuo, and the residue taken up in acetone and filtered through Celite. After concentration, the residue was chromatographed (initial eluant $\mathrm{CHCl}_{3}$; final eluant $2 \% \mathrm{MeOH}$ in $\mathrm{CHCl}_{3}$ ) to give the alcohol 47 ( $155 \mathrm{mg}, 20 \%$ ): $[\alpha]_{\mathrm{D}}+30.2^{\circ}(c 0.44)$; IR $3480,1735 \mathrm{~cm}^{-1}$; MS, $m / z 442.2124$ (442.2077) $\left(\mathrm{MNH}_{4}{ }^{+}\right), 425\left(\mathrm{MH}^{+}\right)$. Also obtained was $4 \beta, 8 \alpha$-diacetoxy- $3 \alpha, 15$-dihydroxy-12,13-epoxytrichothec-9-ene (48) $(151 \mathrm{mg}, 18 \%):[\alpha]_{\mathrm{D}}+17.4^{\circ}(c 0.41)$; IR $3500,1730 \mathrm{~cm}^{-1}$; MS, $m / z 400.1992$ (400.1971) $\left(\mathrm{MNH}_{4}{ }^{+}\right)$.

Cytotoxicity Assay. Cultures of mouse lymphoma cells (L5178Y) were prepared from frozen stocks derived from an original culture supplied by the Central Toxicology Laboratory, ICI (Alderley Edge). Eagle's Minimum Essential Medium supplemented with $10 \%$ or $3 \%$ horse serum (10:90 HS-EMEM, $3: 97$ HS-EMEM) and containing sodium pyruvate ( $200 \mu \mathrm{~g} / \mathrm{mL}$ ) was used for routine subculture and cytotoxicity experiments, respectively. Cells were cultured in suspension in 1-oz polystyrene Universal bottles ( 20 mL per bottle) at $37^{\circ} \mathrm{C}$ with intermittent agitation. Cells selected for use in cytotoxicity experiments were subcultured 24 h previously to ensure that immediately prior to use in an experiment the cells were in log phase growth and at a final cell density not exceeding $5 \times 10^{5}$ cells $/ \mathrm{mL}$. All media contained penicillin ( 100 units $/ \mathrm{mL}$ ), streptomycin ( $100 \mu \mathrm{~g} / \mathrm{mL}$ ), and amphotericin B $(2.5 \mu \mathrm{~g} / \mathrm{mL})$.

Stock solutions of the compounds were made up in DMSO at initial concentrations of $1 \times 10^{-2} \mathrm{~g} / \mathrm{mL}$. Tenfold serial dilutions were then made in DMSO to achieve concentrations down to 1 $\times 10^{-7} \mathrm{~g} / \mathrm{mL}$. These solutions ( $100 \mu \mathrm{~L}$ ) were then diluted 200 -fold with 10:90 HS-EMEM ( 20 mL ) to achieve final concentrations in the culture media of $\left(5 \times 10^{-5}\right)-\left(5 \times 10^{-10}\right) \mathrm{g} / \mathrm{mL}$. L5178Y cells in $\log$-phase growth were seeded at $2 \times 10^{5}$ cells per 20 mL of culture medium containing the test compound or solvent control and grown up in suspension culture for 3 days. At the end of the
growth period, final cell numbers were determined by plating efficiency tested on washed cells from each culture. The cells were suspended in 5 mL of $3: 97$ HS-EMEM (phenol red free), a further 1 in 10 dilution of the cell suspension was made in the same medium, and the crude cell concentration was determined from the standard absorbance curve for L5178Y (wavelength 540 nm ). A nominal $1 \times 10^{5}$ cells were then transferred to 10 mL of fresh 3:97 HS-EMEM, from which 0.1 mL (ca. 1000 cells) was transferred to 23 mL of prewarmed $\left(37^{\circ} \mathrm{C}\right)$ 20:80 HS-EMEM. Molten $5 \%$ Noble Agar ( 1.6 mL ) was added to the dilute cell suspension, which was then mixed vigorously, poured into a $90-\mathrm{mm}$ plastic culture dish (one dish per original treated cell suspension), and allowed to gel for 15 min on an ice-cooled tray. The soft agar cultures were then incubated at $37^{\circ} \mathrm{C}$ in a humidified incubator gassed with $5 \% \mathrm{CO}_{2}$ in air. The cell colonies were counted at the end of a 10 -day incubation period by using an automated colony counter (System III Image Analyser, Micromeasurements Ltd.). The total viable cell yield from each original culture was calculated on the basis of the final plating efficiency in agar, and the cytotoxicity was expressed as a percentage relative to the cell yield in solvent controls. $\mathrm{LC}_{50}$ values were interpolated from the dose-response data for each compound. Since 10 -fold serial dilutions were used in order to cover a wide range of activities, differences $<0.25$ log are not significant. T-2 was used as a positive control with each batch of determinations.

Registry No. $1,21259-20-1 ; 1$ ( $\mathrm{R}_{1}=\mathrm{OC}(\mathrm{S})$ imidazole), 118143-05-8; 2, 2270-40-8; 3, 36519-25-2; 4, 21259-21-2; 5, 63148-31-2; 6, 115589-76-9; 7, 118142-89-5; 8, 118142-90-8; 9, 118204-86-7; 10, 68165-53-7; 11, 65041-92-1; 12, 65725-06-6; 13, 99127-69-2; 14, 118142-91-9; 15, 74516-69-1; 16, 118142-92-0; 17, 26934-87-2; 18, 34114-98-2; 19, 34114-99-3; 20, 34084-03-2; 21, 118142-93-1; 22, 118142-94-2; 23, 118204-87-8; 24, 118142-95-3; 25, 118142-96-4; 26, 118142-97-5; 27, 118142-98-6; 28, 118142-99-7; 29, 118143-00-3; 30, 113706-92-6; $30\left(\mathrm{R}_{1}=\mathrm{OH}, \mathrm{R}_{6}=\right.$ OTBDMS), 118143-06-9; $30\left(\mathrm{R}_{6}=\right.$ OTBDMS $), 113728-56-6 ; 30\left(\mathrm{R}_{6}=\mathrm{OAc}\right)$, 118143-07-0; $30\left(\mathrm{R}_{6}=\right.$ OCOEt $), 118169-52-1 ; 30\left(\mathrm{R}_{6}=\right.$ OCOPr $)$, 118170-26-6; $30\left(\mathrm{R}_{6}=\mathrm{OCOBu}\right), 118143-08-1 ; 30\left(\mathrm{R}_{6}=\right.$ OCOPe $)$, $118143-09-2 ; 30\left(\mathrm{R}_{6}=\mathrm{OCOPh}\right), 118143-10-5 ; 31,111112-47-1 ; 32$, 98813-18-4; 33, 116163-74-7; 34, 118143-01-4; 35, 118143-02-5; 36, 77620-47-4; 37, 77620-53-2; 9 9 -38, 118243-15-5; 9ß-38, 118204-88-9; $9 \alpha-39,118243-16-6$; 9ß-39, 118204-89-0; 40, 118204-90-3; 41, 118170-22-2; 42, 118170-23-3; 43, 118143-03-6; 44, 76348-84-0; 45, 118170-24-4; 46, 118170-25-5; 47, 118143-04-7; 48, 65180-29-2.

# Long Acting Dihydropyridine Calcium Antagonists. 2. 2-[2-Aminoheterocycloethoxy]methyl Derivatives 

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A series of [(2-aminoheterocycloethoxy)methyl]dihydropyridines were prepared as selective coronary vasodilators. Results showed that a wide variety of five- and six-membered heterocycles were acceptable at the 2 -position of the dihydropyridine ring and in vitro potency and tissue selectivity was independent of the basicity of these heterocycles. The SAR indicated that activity was optimum when the largest ester group was placed at the 3 rather than 5 position. 2-[[2-[(3-Amino-1 H -1,2,4-triazol-5-yl)amino]ethoxy]methyl]-4-(2,3-dichlorophenyl)-3-(ethoxycarbonyl)-5-(meth-oxycarbonyl)-6-methyl-1,4-dihydropyridine ( 3 b ) (UK-52,831) emerged as a potent ( $\mathrm{IC}_{50}=6.3 \times 10^{-9} \mathrm{M}$ ) and tis-sue-selective calcium channel blocker with a duration of action $>7 \mathrm{~h}$ in the anaesthetized dog.

In an earlier publication, ${ }^{1}$ we reported the preparation and structure-activity relationship (SAR) of a series of novel dihydropyridine (DHP) calcium antagonists substituted in the 2 -position by various basic side chains. The aim of our work was to modify the physicochemical

[^8]properties of the DHP system such that bioavailability and duration of action were superior to currently available agents. These studies led to the identification of amlodipine (1a), which is presently in late stage clinical de-

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